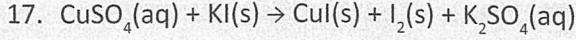
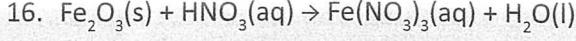
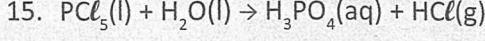
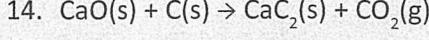
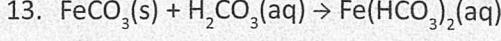
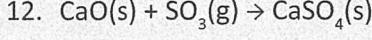
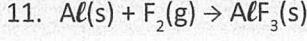
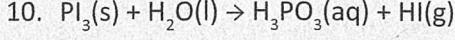
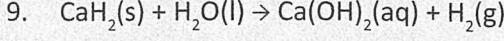
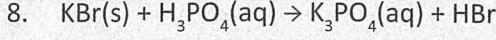
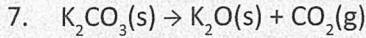
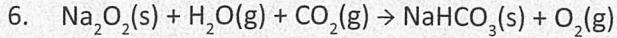
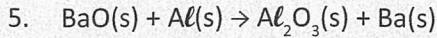
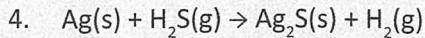
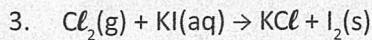
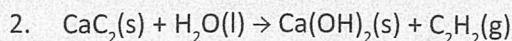
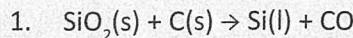
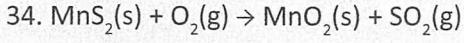
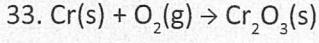
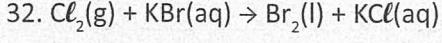
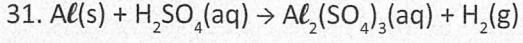
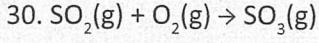
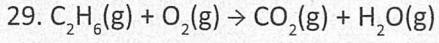
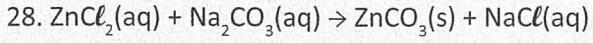
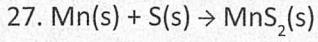
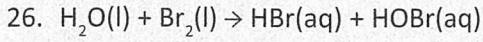
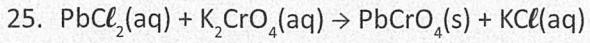
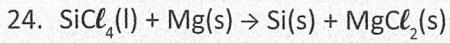
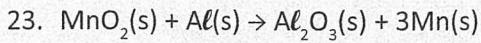
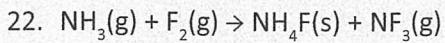
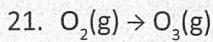
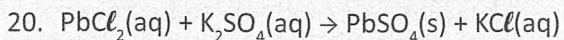
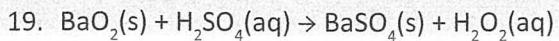
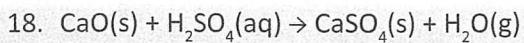


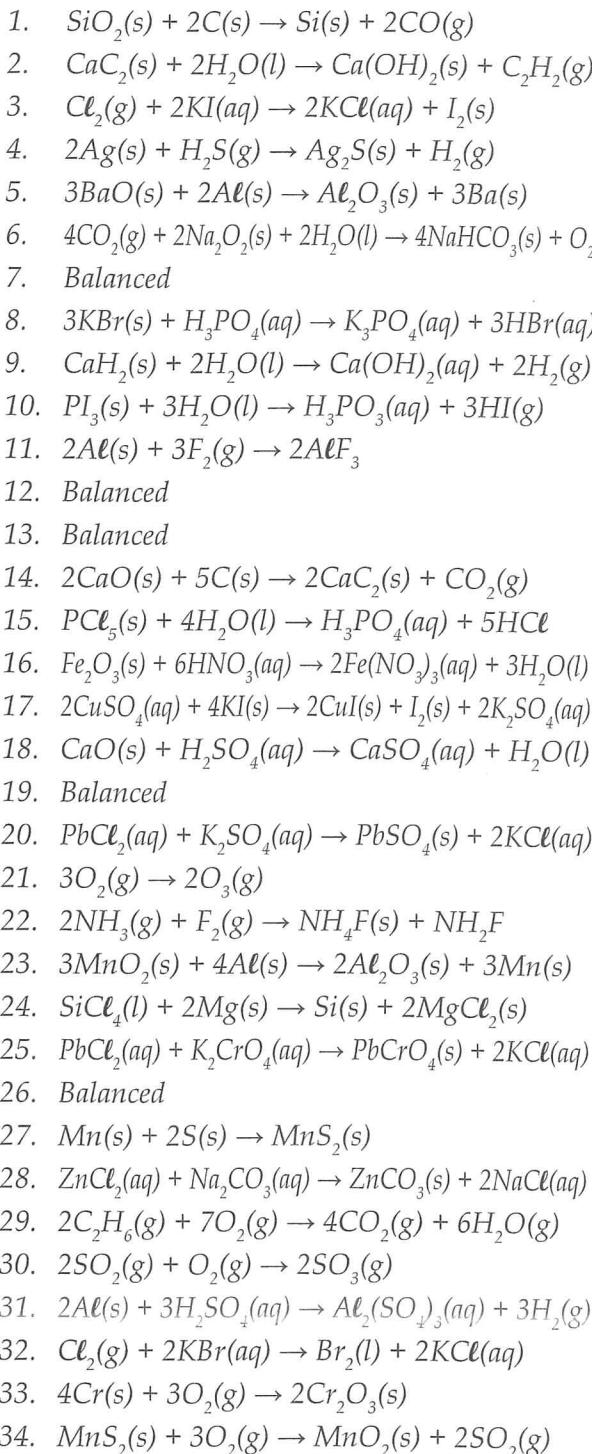
II. Balancing Chemical Equations

Balance each of the following chemical equations:





II.



Multiple Choice Questions

1. d, 2. c, 3. b, 4. c, 5. c, 6. b, 7. d, 8. c, 9. d.
10. d

Chapter 6. **Electron Configuration and the** **Periodic Table**

1. Valence electrons are found in the outer-most shells. They determine the valence of the elements and hence, their chemical behaviour.